

# A Review of Thermal Energy Storage Systems with Salt Hydrate Phase Change Materials for Comfort Cooling

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# Table of Content

- Background
- Thermal Energy Storage System Performance
  - Phase Separation
  - Subcooling
  - Encapsulation
  - Heat Transfer Enhancement
  - Experimental and Numerical System Studies
- Conclusion



# Background

- Latent heat thermal energy storage (LHTES) with phase change materials (PCMs) : high energy storage density and small temperature change.
  - District cooling network
    - additional cooling power
  - Chiller based system
    - alleviate peak grid electricity
    - lower marginal electricity cost
    - Increase production efficiency
  - Free Cooling storage
    - night time storage, daytime use



# Methodology

- Salt hydrate based cold thermal energy storage systems for comfort cooling.
- Review of 100 papers focusing on enhancement of TES:
  - Phase Separation
  - Subcooling
  - Encapsulation
  - Heat Transfer Enhancement
  - Experimental and Numerical System Studies



# Phase Change Materials

Table 1 Comparison of Organic with Inorganic PCMs

	Organic	Inorganic	Eutectic
Pros	<ul style="list-style-type: none"> <li>• Low Cost (120Euro/kWh)</li> <li>• Self nucleating</li> <li>• Chemically inert and stable</li> <li>• No phase segregation</li> <li>• Recyclable</li> <li>• Available in large temperature range</li> </ul>	<ul style="list-style-type: none"> <li>• Moderate cost (130 Euro/kWh)</li> <li>• High volumetric storage density (180-300 MJ/m<sup>3</sup>)</li> <li>• Higher thermal conductivity (0.6W/mK)</li> <li>• Non flammable</li> <li>• Low volume change</li> </ul>	<ul style="list-style-type: none"> <li>• Sharp melting point</li> <li>• Low volumetric storage density</li> </ul>
Cons	<ul style="list-style-type: none"> <li>• Flammable</li> <li>• Low thermal conductivity (0.2W/mK)</li> <li>• Low volumetric storage density (90-200 MJ/m<sup>3</sup>)</li> </ul>	<ul style="list-style-type: none"> <li>• Subcooling</li> <li>• Phase segregation</li> <li>• Corrosion of containment material</li> </ul>	<ul style="list-style-type: none"> <li>• Limited availability</li> </ul>

# Phase Change Materials

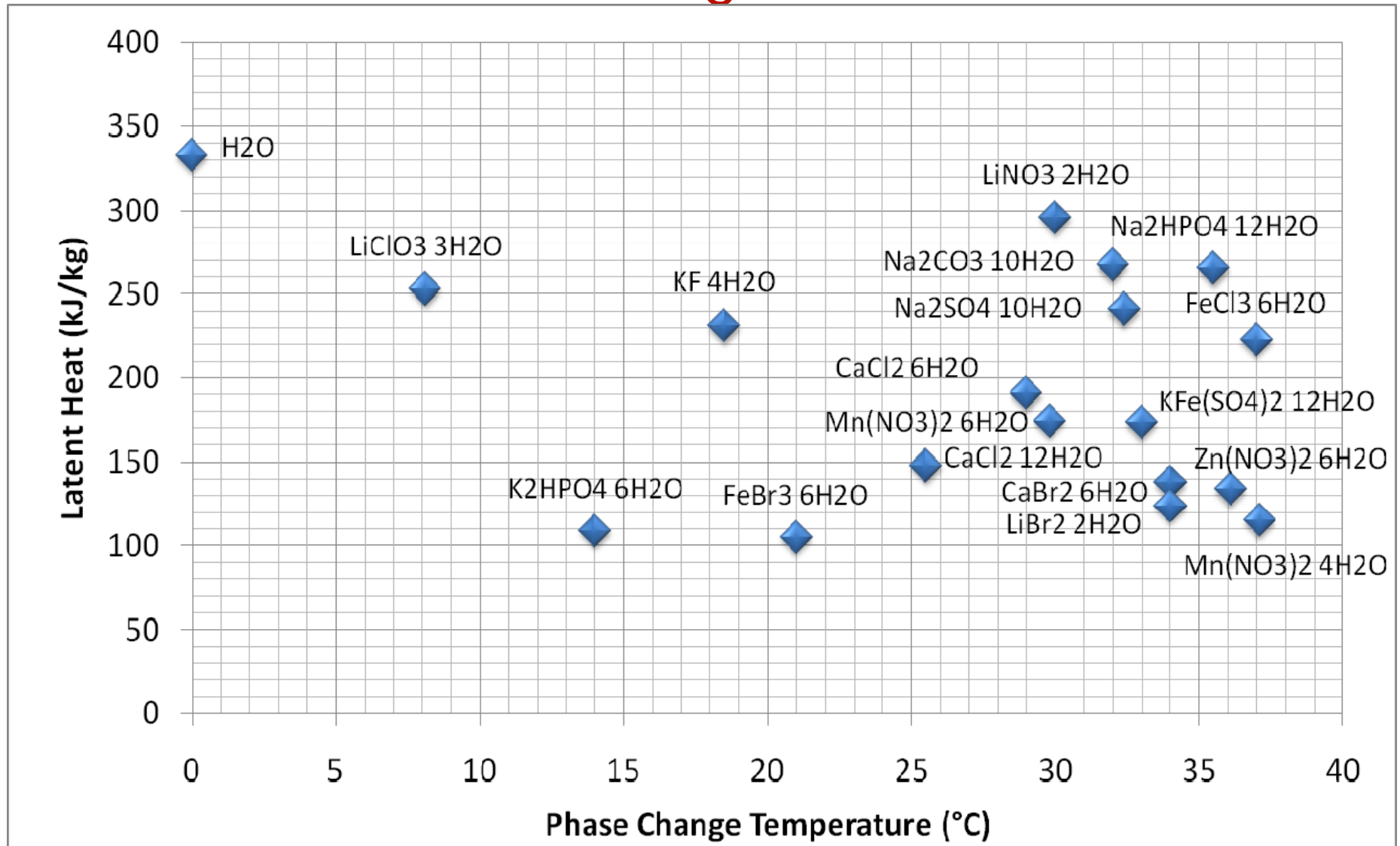


Figure 1 Analytical Grade Salt Hydrates

# Phase Change Materials

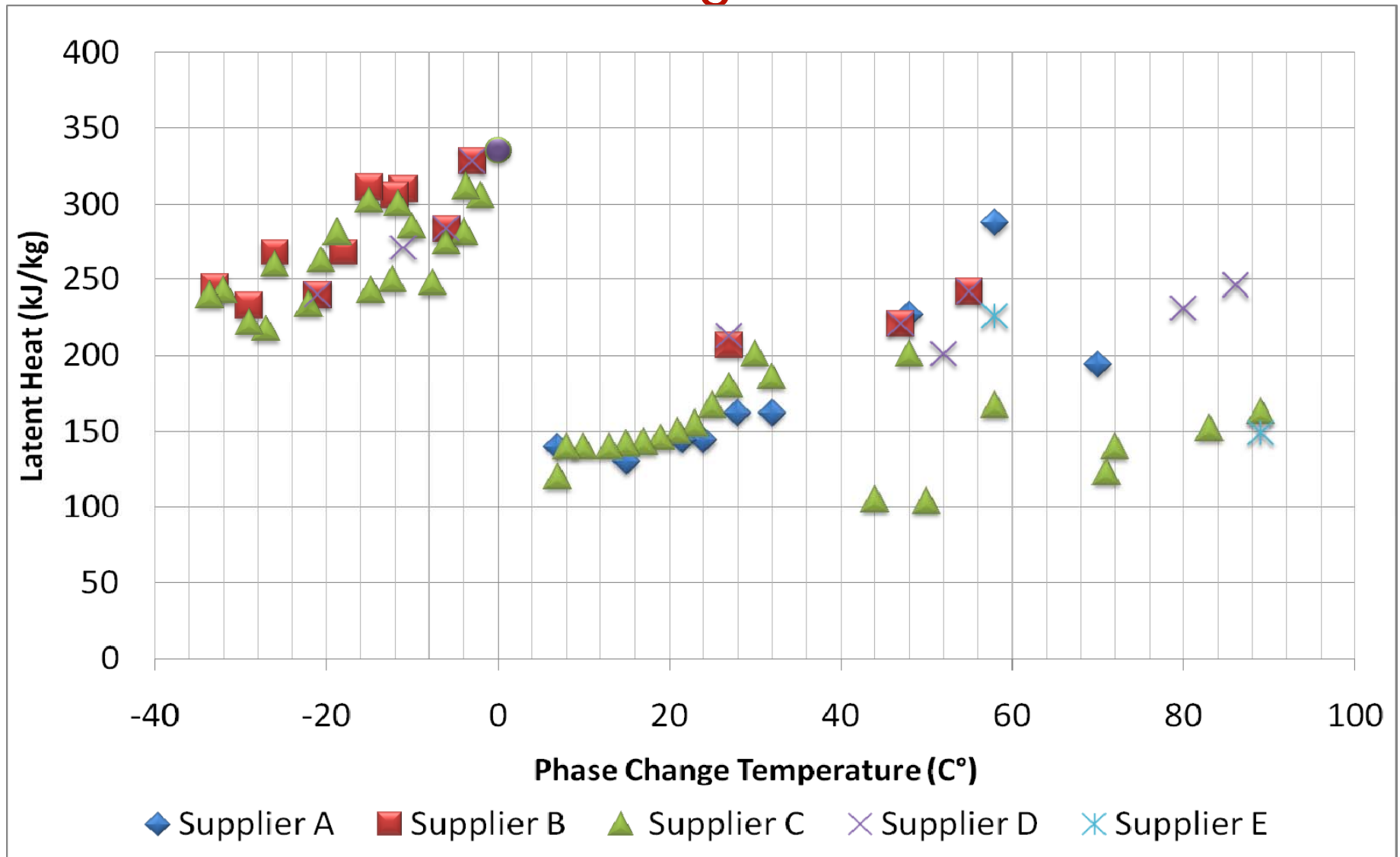
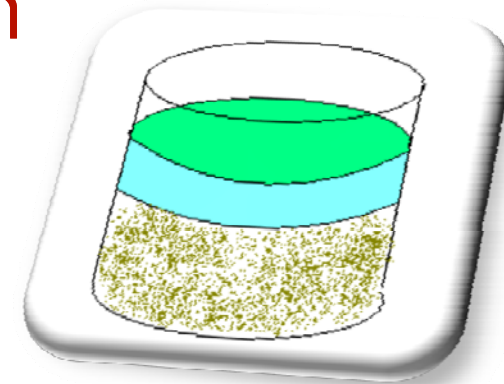


Figure 2 Commercialized Salt Hydrate Products

# Thermal Energy Storage System Performance

## -Phase Separation



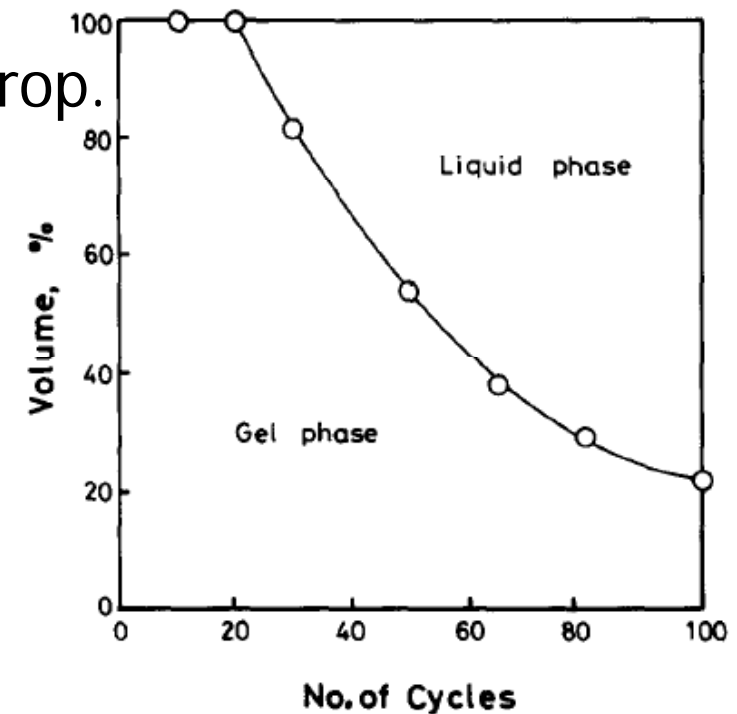
### • Incongruent Melting

→ Gelling agents: bentonite, starch, cellulose, super absorbent copolymer, carboxymethyl cellulose, etc.

- 20% ~ 35% latent enthalpy drop.

Artificial mixing heat storage system

- parasitic electricity load



Ryu et al. 1992



# Thermal Energy Storage System Performance -Subcooling

→ Cold finger, porous heat exchange surface, and nucleating agents: carbon nanofibers, copper, titanium oxide, potassium sulfate, borax, etc.

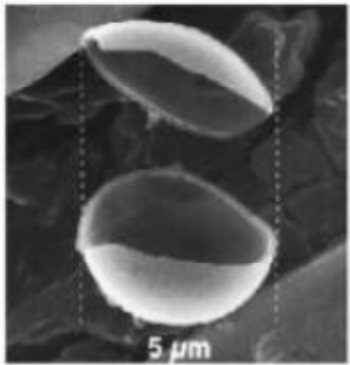
Supercooling range of the thickened PCMs with the respective nucleating agents

PCM	Thickener	$T_m$ (°C)	Nucleating agent (size, $\mu\text{m}$ )	Supercooling (°C)	
				w/o nucleator	w/nucleator
$\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$	SAP	32	Borax ( $20 \times 50 - 200 \times 250$ )	15-18	3 -4
$\text{Na}_2\text{HPO}_4 \cdot 12\text{H}_2\text{O}$	SAP	36	Borax ( $20 \times 50 - 200 \times 250$ )	20	6 -9
			Carbon (1.5-6.7)		0 -1
			$\text{TiO}_2$ (2-200)		0 -1
			Copper (1.5-2.5)		0.5-1
			Aluminum (8.5-20)		3 -10
$\text{CH}_3\text{COONa} \cdot 3\text{H}_2\text{O}$	CMC	46	$\text{Na}_2\text{SO}_4$	20	4 -6
			$\text{SrSO}_4$		0 -2
			Carbon (1.5-6.7)		4 -7
$\text{Na}_2\text{S}_2\text{O}_3 \cdot 5\text{H}_2\text{O}$	CMC	57	$\text{K}_2\text{SO}_4$	30	0 -3
			$\text{Na}_2\text{P}_2\text{O}_7 \cdot 10\text{H}_2\text{O}$		0 -2

Ryu et al. 1992

• Subcooling reduced from  $20^\circ\text{C}$  to  $2^\circ\text{C}$ ,  $2^\circ\text{C}$  = high efficiency loss.

# Thermal Energy Storage System Performance -Encapsulation



BASF



Serve as heat transfer surface, prevents PCM from reacting, and adds mechanical strength.

- **Macro encapsulation:** easy handling, but low IPF

$$H_{\text{spherical capsules}} > H_{\text{cylindrical}} > H_{\text{plate type}} > H_{\text{tubular capsules}}$$

- **Micro encapsulation:** prevents phase separation, but has high production cost

- **Bulk storage:** no packaging cost and high storage density.



EPS

# Thermal Energy Storage System Performance

## -Heat Transfer Enhancement

- Fins placed in the same direction as tubes for vertical tube setup:  $90\text{W/m}^2\text{K}$  and  $250\text{W/m}^2\text{K}$
- Dispersion of aluminum and graphite
  - $1.6\text{X}$  to  $20\text{X}$
- Impregnation of PCM into a graphite matrix
  - $20\text{X}$  to  $1000\text{X}$
- Today: **organic compounds**
  - Tomorrow: **salt hydrates**



# Thermal Energy Storage System Performance

## -Experimental and Numerical System Studies

- Space thickness, HTF entry temperature, encapsulation conductivity and solid phase PCM conductivity have significant influence on melting/ solidifying process.
- Passive walls with PCM : better performance than masonry wall of 5X thickness.
- Case study in Saudi Arabia: 23% to 40% cooling power reduction.
- Case study of office building in Stockholm: 5% to 30% reduction.
- Peak cooling demand may be reduced by 40% to 90% with **proactive control** and weather forecasts.



# Conclusions

- Techno-economical PCM systems: lower cost, higher power, and larger storage density
  - advanced material research: low subcooling, phase separation
  - system performance modeling, design optimization and experimental work
- Lack of accurate commercial PCM property → discrepancies between design model and actual TES system.
- Move towards standardization of property measurement.
- ❑ Salt hydrate based TES seems to be one of the most promising technologies for integration in the built environment.



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# Thank you

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